



Full Length Research Article

ACCUMULATION OF SOME HEAVY METALS Cd, Cu, Pb AND Zn IN MUGIL CEPHALUS,
PERNA VIRIDIS AND PENAEUS MONODON

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ARTICLE INFO

Article History:

Received 2nd January, 2010
Received in revised form
28th February, 2011
Accepted 21st March, 2011
Published online 13th April, 2011

Key words:

Bioconcentration factor, cadmium, copper, lead, zinc, *Mugil cephalus*, *perna viridis* and *Penaeus monodon*.

ABSTRACT

In the present study *Mugil cephalus*, *Perna viridis* and *Penaeus monodon* was exposed to cadmium, copper, lead and zinc under long term chronic toxicity test to investigate the bioaccumulation pattern of cadmium, copper, lead and zinc to the three marine test organisms. Treated test organisms under long term chronic toxicity test showed that the concentration of heavy metals in the tissues were highly significant ($P < 0.001$) when compared with control. The order of concentration in the tissues of *M. cephalus* was $Zn > Cu > Pb > Cd$, *P. viridis* accumulated in the order of $Cd < Pb < Cu < Zn$ and *Penaeus monodon* accumulated heavy metals in the order of $Cu > Zn > Pb > Cd$. Knowledge on accumulation and distribution of metals in the soft tissues may help us to understand the processes involved in the uptake and excretion of metals. More studies should be conducted in the future to determine the potentials of *Mugil cephalus*, *Perna viridis* and *Penaeus monodon* as biomonitoring agents in heavy metal pollution.

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INTRODUCTION

Aquatic pollution started long back but intensified during the last few decades, and now the situation has become alarming, especially in India (Girija *et al.*, 2007). Environmental pollution in urbanized countries have deteriorated the quality of the water (Zhao *et al.*, 2011). Contamination of aquatic ecosystems with heavy metals has been receiving increased worldwide concern (Tsangaris *et al.*, 2007). Copper enters the aquatic environment chiefly through leaching from paints on the hulls of boats and ships (Singh and Turner, 2009). Due to persistence in the environment and tendency to accumulate in the biota, copper pose a potential hazard to environmental and human health. It is the most poisonous heavy metal when present in excess (De Boeck *et al.*, 2006). Zinc is the fourth most widely used metal in the world. Its major uses include galvanized steel for alloy production, and as an ingredient in rubber and paints (USEPA, 1991). Cadmium is toxic to fish even at low concentrations (Jarup and Akesson, 2009). Due to its long biological half-life and strong ability to accumulate in animal tissues, residual cadmium forms a serious threat to the performance and survival of aquatic biota (Seebaugh *et al.*, 2005). Sources of lead in marine environments include natural sources from rock weathering, riverbank and coastal erosion, and anthropogenic sources from urban and industrial

emissions (Kelly *et al.*, 2009). The impacts vary relatively minor to major disruptions due to bioaccumulation and biomagnification processes (Altun *et al.*, 2008). The ecological integrity is judged using toxicity tests (Tueros *et al.*, 2009). The toxicity tests measure the integrated responses to the possible acute or chronic effects of contaminants, on these processes (Watts and Pascoe, 2000). Test species should be sensitive enough to respond to low levels of contaminants and must be available for use from field collection throughout the year. Chronic toxicity tests data are generally more reliable, providing responses related to a complete or part of life cycle of the test-species (Nascimento *et al.*, 2000). Aquatic organisms exposed to a higher concentration of trace metals in water may take up substantial quantities of these metals (Kord *et al.*, 2010). Accumulation of heavy metals in tissues mainly depends upon concentration of metals in water and exposure period; although some other environmental factors such as salinity, pH, hardness and temperature play significant roles in metal accumulation (Blackmore and Wang, 2003). Marine organisms are characterized by a greater spatial ability to accumulate some metals when compared with bottom sediments (Kaladharan *et al.*, 2005). Mussels have been considered as a potential biomonitor for metallic contamination in marine ecosystems (Jung and Zauke, 2008), Fish are the good bioindicator for monitoring metal pollution (Tyrrell *et al.*, 2005). Species of prawn can be used to monitor the trace element pollution in the aquatic environment because

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they are omnivorous benthic animals that maintain their body in direct contact with the water and substrate of their environment and they tend to accumulate metals in their tissues (Barrento *et al.*, 2009). Hence in the present study the *Mugil cephalus*, *Perna viridis* and *Penaeus monodon* was exposed to cadmium, copper, lead and zinc under long term chronic toxicity test for 30 days to estimate the bioconcentration factors of cadmium, copper, lead and zinc.

MATERIALS AND METHODS

Collected fingerlings of *Mugil cephalus* of mean 1.5 ± 0.4 cm in length and 0.13 ± 0.02 g in weight, juvenile specimens of *Perna viridis* (1.6 ± 0.4 cm in length and 0.12 ± 0.01 g) and post-larval stages of *Penaeus monodon* (PL-12) were immediately transported to the laboratory in air filled plastic bags and acclimatized in glass aquaria and fish fingerlings in 200 L Fiberglass Reinforced Plastics (FRP) tanks with aerated natural filtered seawater for a period of 8 days at 28 PSU salinity, temperature of 28 ± 2 °C, dissolved oxygen of 5.6 mg/l and pH of 8.01. Captured wild organisms were quarantined immediately (Oxytetracycline). After a day of acclimatization, the fry specimens of *M.cephalus* were then fed with pellets of rice bran and oil cake, *P.viridis* was fed with mass culture of cyanobacteria (*Anabaena* sp.) Samples of cyanobacteria were isolation was done using serial dilution and streaking plate method (Rippka, 1988). Samples were diluted with sterilized ASN-III medium up to 10-25 dilution. Dilution tubes were incubated under constant light at room temperature of 28 ± 3 °C. Stock cultures were maintained at room temperature under diffused light. Post larvae of *Penaeus monodon* were fed with mixed feed for *P.monodon* (Japan) throughout acclimatization period. The dead animals were removed immediately. The remaining detritus were removed by siphoning (USEPA, 1996).

Prior to toxicity tests and stock solution preparations, all the glasswares were washed in 10 per cent nitric acid and rinsed with deionized water. Stock solutions of cadmium, copper, lead and zinc were freshly prepared by dissolving the proper metal salts ($\text{CdCl}_2 \cdot 2.5\text{H}_2\text{O}$ for Cd, CuCl_2 for Cu, $\text{Pb}(\text{NO}_3)_2$ for Pb and $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$ for Zn in deionized (double distilled) water with glass standard flasks. Stock solutions were acidified by the addition of 0.1 ml of concentrated nitric acid per litre of stock solution (Chapman, 1978). Fresh stock solutions were prepared daily. These solutions were serially diluted to get the experimental concentration for the toxicity test. The experimental method includes static renewal (24 hour renewal) test by following the method of USEPA (2002a). Five concentrations in a geometric series including control were prepared for the test for 30 days for short-term chronic toxicity test (USEPA, 2002b). Toxicant and seawater were replaced on daily basis. Dilution water for the experiment was collected from the unpolluted site (Neelangarai, Tamilnadu, India) and filtered through $0.45\mu\text{m}$ filter paper (HA-Millipore) using Millipore vacuum pump. Each series of test chambers consisted of duplicates with 10 animals in a 5 L glass trough. Test chambers were loosely covered to reduce evaporation and to minimize the entry of dust into solutions and to prevent loss of test animals. All the experiments were conducted at salinity of 28 PSU, temperature of 28 ± 2 °C, dissolved oxygen of 5.6 mg/l and pH of 8.01 with gentle aeration. Test animals were fed regularly three times a day. Temperature, pH, salinity,

dissolved oxygen and test concentrations were measured to ensure the acceptability and validation of the tests, following standard methods (USEPA, 1996). Daily observations were recorded for survival and mortality. The criterion for determining death was the absence of movement when the animals were gently stimulated. Dead animals were removed at each observation and survivors were counted (USEPA, 2002b). The metal content of the tissue samples was based on dry weight. The organisms collected from the test chambers of the chronic test were sacrificed and was stored to -4 °C and then transported to the laboratory and stored at -20 °C in the deep freezer until analysis. During the course of analysis, the tissue was washed with distilled water and dried at 95 °C in hot air oven and grinded to a fine powder with pestle and mortar and the metal analysis was carried out by UNEP (1984). To ensure the accuracy and precision in the sample analysis, it was cross-examined with respect to certified reference material (DOLT-3, Dogfish liver certified reference material for trace metal, from national research council Canada) (DOLT-3, 1999) (Table 1). Nearest gram of the dried tissue powder was transferred to a Teflon crucible. To the tissue added 8-10 ml of concentrated acid (60 per cent nitric acid (HNO_3); 70 per cent perchloric acid (HClO_4)), such that the tissue was totally wet and with slight excess of acid and left it at room temperature for 12 h. The digested samples were heated slowly to 180 °C on hot plate, till the sample volume was reduced to 2-3 ml. The resulting colourless solution was made up to 25 ml in standard glass flask and stored in 50 ml Polyethylene-Terephthalate (PET) bottles and was analysed for metals in Varian SpectraAA 220FS Atomic absorption spectrophotometer. Suitable internal chemical standards (Merck Chemicals, Germany) were used to calibrate the instrument. All the reagents used were analytical grade of high purity. The results were expressed as $\mu\text{g/g}$ dry weight.

RESULTS

Treated heavy metals *M. cephalus* for short-term chronic toxicity test showed that the concentration of heavy metals in the tissues were highly significant ($P < 0.001$) when compared with control (One way ANOVA: Dunnetts multiple comparison test ($\alpha = 0.05$)). Zinc concentrations in the tissues were higher than copper. The order of concentration in the tissues of *M. cephalus* was $\text{Zn} > \text{Cu} > \text{Pb} > \text{Cd}$ (Table 2). Values were highly significant ($P < 0.001$) when compared with control in the test conducted (One way ANOVA: Dunnetts multiple comparison test ($\alpha = 0.05$)). The order of decreasing concentration in the tissue of *P.viridis* was $\text{Cd} < \text{Pb} < \text{Cu} < \text{Zn}$ (Table 3). Post larvae of *P.monodon* exposed to cadmium, copper, lead and zinc in the short-term chronic toxicity test revealed that the copper was concentrated more than zinc. Values were highly significant ($P < 0.001$) when compared with control in the test conducted (One way ANOVA: Dunnetts multiple comparison test ($\alpha = 0.05$)). The order of concentration in the tissue was $\text{Cu} > \text{Zn} > \text{Pb} > \text{Cd}$ (Table 4).

DISCUSSION

In the long term chronic toxicity test conducted for 30 days showed that the accumulation of heavy metals and the essential metals such as copper and zinc in the tissues of *M.cephalus* was high when compared with cadmium and lead which had significant ($P < 0.0001$) increase with control. The

Table 1. Recovery of trace elements in certified reference material (DOLT-3) Dogfish liver certified reference material for trace metals

Element	Certified values (mg/kg)	Measured concentration (mg/kg)	Recovery (%)
Cd	19.4 ±0.6	20.03 ±0.11	103.26
Cu	31.2 ±1.0	31.7 ±0.63	101.45
Pb	0.319 ±0.045	0.311 ±0.05	97.19
Zn	86.6 ±2.4	85.62 ±3.22	98.87

*Measure concentration (mg/kg) is the mean and standard deviation of n=12

Table 2. Concentrations of cadmium, copper, lead and zinc in the tissues of *M.cephalus* exposed under short-term chronic toxicity test

Metal	Units	Concentrations (mg/l)					
Cd	a	0	10	20	40	80	160
	b	0.43 ±0.04	7.05 ±0.14	9.13 ±0.32	12.15 ±0.78	15.50 ±0.07	26.95 ±0.35
Cu	a	0	10	20	40	80	160
	b	4.20 ±0.21	9.78 ±0.46	22.55 ±0.21	43.35 ±0.78	79.98 ±1.66	147.20 ±2.26
Pb	a	0	51	76	114	171	256
	b	0.43 ±0.35	4.68 ±0.11	8.35 ±0.78	11.98 ±0.39	21.60 ±0.78	35.03 ±1.59
Zn	a	0	29	46	74	118	188
	b	11.93 ±0.32	20.75 ±1.20	44.10 ±1.06	78.35 ±7.85	127.75 ±7.00	241.55 ±3.96

Values were significant at $P<0.05$, One way ANOVA: Dunnetts multiple comparison ($\alpha=0.05$), $P<0.05$ values compared with control were highly significant (***, $P<0.001$). a, exposed concentration (mg/l), b, concentration in tissues ($\mu\text{g/g}$ dry wt.); Values are mean and standard deviation each n=2; The concentration column (mg/l) contains '0' indicating control in the test conducted in triplicate

Table 3. Concentrations of cadmium, copper, lead and zinc in the tissues of *P.viridis* exposed under short-term chronic toxicity test

Metal	Units	Concentrations (mg/l)					
Cd	a	0	16	26	41	66	105
	b	1.65 ±0.07	3.78 ±0.35	8.00 ±0.07	11.43 ±0.39	20.50 ±0.14	38.80 ±0.78
Cu	a	0	10	15	23	34	51
	b	1.90 ±0.07	3.80 ±0.01	7.95 ±0.56	12.40 ±0.21	25.30 ±0.35	52.90 ±4.59
Pb	a	0	10	15	23	34	51
	b	1.73 ±0.35	3.65 ±0.14	7.78 ±0.11	15.83 ±0.32	31.40 ±0.42	42.43 ±0.39
Zn	a	0	8	16	32	64	128
	b	11.93 ±0.32	13.25 ±0.78	21.88 ±0.39	40.03 ±0.81	70.75 ±1.20	113.75 ±3.25

Values were significant at $P<0.05$, One way ANOVA: Dunnetts multiple comparison ($\alpha=0.05$), $P<0.05$ values compared with control were highly significant (***, $P<0.001$). a, exposed concentration (mg/l), b, concentration in tissues ($\mu\text{g/g}$ dry wt.); Values are mean and standard deviation each n=2; The concentration column (mg/l) contains '0' indicating control in the test conducted in triplicate

Table 4. Concentrations of cadmium, copper, lead and zinc in the tissues of *P.monodon* exposed under short-term chronic toxicity test

Metal	Units	Concentrations (mg/l)					
Cd	a	0	10	15	20	25	30
	b	1.28 ±0.35	2.83 ±0.35	5.90 ±0.34	12.43 ±0.39	22.60 ±0.28	35.83 ±0.32
Cu	a	0	8	13	21	34	54
	b	4.55 ±0.14	10.08 ±0.21	22.55 ±0.67	43.35 ±0.78	79.98 ±1.66	153.33 ±3.22
Pb	a	0	2	4	8	16	32
	b	2.98 ±0.39	4.03 ±0.11	7.78 ±0.11	15.83 ±0.32	31.40 ±0.42	44.65 ±0.35
Zn	a	0	15	23	34	51	76
	b	16.60 ±0.78	16.10 ±0.64	27.75 ±0.71	40.03 ±0.81	70.75 ±1.20	108.60 ±2.83

Values were significant at $P<0.05$, One way ANOVA: Dunnetts multiple comparison ($\alpha=0.05$), $P<0.05$ values compared with control were highly significant (***, $P<0.001$). a, exposed concentration (mg/l), b, concentration in tissues ($\mu\text{g/g}$ dry wt.); Values are mean and standard deviation each n=2; The concentration column (mg/l) contains '0' indicating control in the test conducted in triplicate

deficiency of zinc can provoke serious consequences, as decrease of growth and sexual immaturity (Ansari *et al.*, 2004). The excess of some metals can cause harmful effects in fish such as alterations in oxygen consumption and damages in the gills (Zagatto and Bertolotti, 2006). Copper exposures leads to increased ammonium production (Kunwar *et al.*, 2009). This may be the reason for the reduced swimming performance of copper exposed fish, and could be caused by an increase in the catabolism of proteins (Waser *et al.*, 2009). It is well known that heavy metals accumulated in substantially high levels can be very toxic for fish, especially for young and eggs which are very sensitive to pollution (Kalay and Erdem, 1995).

Usero *et al.* (2004) reported in *Liza aurata*, *Anguilla anguilla* and *Solea vulgaris* the priority of metal concentrated in the tissues were generally $\text{Cu} = \text{Zn} > \text{Cd}$. Metal levels in *Diplodus annularis*, *Scorpaena porcus* and *S.scrofa* in the order $\text{Zn} > \text{Cu} > \text{Cd}$ (Chaffai *et al.*, 1995). Besides concentrations ($\text{Zn} > \text{Cu} > \text{Cd}$), Chen *et al.* (2004) reported metal concentrations in *Liza macrolepis*, with descending order similar to results of the present study ($\text{Zn} > \text{Cu} > \text{Cd}$). In the present study *M.cephalus* accumulated heavy metals in the tissues of an order of $\text{Zn} > \text{Cu} > \text{Pb} > \text{Cd}$, which were comparable with all the above studies. Only levels of zinc and copper were comparable to those reported elsewhere (Tyrrell *et al.*, 2005). Biney and Ameyibor (1992) reported that the accumulation of

copper, lead, zinc and cadmium in pink shrimp (*Penaeus notialis*) were lower than the present study. High concentrations of copper and zinc in both fish and shrimp were similar to findings previously recorded for crustacean (Hossain and Khan, 2001) and fish (Tyrrell *et al.*, 2005). Cadmium and zinc concentrations found in the muscle tissue of *P.monodon* were similar to those found in the shrimps reported by Eisler (1981) and in Pacific shrimps by Harding and Goyette (1989). Copper concentrations were similar to those recorded for brown and rock shrimps from the USA (Texas) continental shelf (Horowitz and Presley, 1977), but intermediate between levels of Australian shrimps *P. merguensis* and *P. monodon* (Darmono and Denton, 1990) and North-east Pacific shrimps *Pandalopsis dispar* and *Pandalus borealis* (Harding and Goyette, 1989).

Dumalagan and Gonzales (2010) reported that among the zinc was the most bioaccumulated in the soft tissues of mussels *P.viridis* followed by copper, than lead. In the present study the *P.viridis* exposed to cadmium, copper, lead and zinc in short-term chronic toxicity test revealed that at the end of the 30th day the metal concentrations were significantly ($P<0.0001$) higher than the green mussels exposed in control. Yap *et al.* (2003) reported high accumulation of zinc in soft mussel tissues of *P. viridis* and suggested that the byssus of mussels is a better biomonitoring organ for zinc contamination. The most abundant trace metals were zinc and copper followed in much lower concentrations by lead and cadmium in clam samples (Das *et al.*, 2009). For bivalves there appear to be trends in metal accumulation patterns of essential trace metals depending on the group, their metabolism and their ability to either excrete or store particular trace elements (Rainbow, 1993). Zinc concentrations of $>2000 \mu\text{g/g}$ dry weight are regularly reported in soft tissues of different oyster species (Hayes *et al.*, 1998). The zinc tissue concentrations determined in the present study for the *P.viridis* were below $113.75 \mu\text{g/g}$ which is more typical less. Copper concentrations in *P.viridis* in the general ranged $3.8\text{-}52.9 \mu\text{g/g}$ can be compared with those for other bivalves. They are similar to concentrations reported for the European cockle *Cerastoderma edule* and mussel *Mytilus edulis* (Hummel *et al.*, 1997). They are also similar to the copper concentrations in the pearl oyster *Pinctada radiata* (Al-Sayed *et al.*, 1994), where the average range was 3.5 and $15.3 \mu\text{g/g}$ from the Arabian Gulf. The copper concentrations in the *P.viridis* were however lower than some other bivalves including *Macoma balthica* and mussels (Hummel *et al.*, 1997). Lead concentrations observed in *P. viridis* in the present study were 2-4 times lower than those reported by Krishnakumar *et al.*, (1998).

Elevated zinc concentration in invertebrates is associated with the presence of a sulphide-transporting protein and with zinc at its active site (Flores *et al.*, 2005), the bivalve gill, for instance, is widely known to lose bioaccumulated metals rapidly when placed in a metal-free environment (Ng and Wang, 2004). Although findings have shown body size to be a critical factor influencing metal concentration in marine bivalves (Chong and Wang, 2001), extremely high concentrations of zinc in green mussels during the whole period of accumulation indicate an important role of mucus in the depuration of metals.

Conclusion

Our study has shown that ability of *Mugil cephalus* to accumulate primarily zinc followed by copper, lead and cadmium, *Perna viridis* accumulated cadmium primarily followed by lead, copper and zinc. *Penaeus monodon* accumulated heavy metals in the order of $\text{Cu}>\text{Zn}>\text{Pb}>\text{Cd}$ exposed under long term chronic toxicity test. High significance was observed in treated ($P<0.001$) when compared with control. This study shows that *Mugil cephalus*, *Perna viridis* and *Penaeus monodon* to be a better biomonitoring agents in heavy metal pollution of environmental monitoring.

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